

Questions (INJSO 2010)

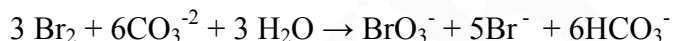
Section A: Questions 1 to 60 are multiple choice with every correct answer carrying 1 mark and every wrong answer carrying -0.25 mark.

SECTION A

Q1. Which of the following organelle is the site for ribosome synthesis?

- a) Ribosomes
- b) Cytoplasm
- c) Nucleus
- d) Nucleolus

Q2. In the chemical reaction



- a) bromine is reduced and water is oxidized
- b) bromine is oxidized and carbonate is reduced
- c) bromine is both oxidized and reduced
- d) bromine is neither oxidized nor reduced

Q3. A flat mirror creates a virtual image of your face. Which of the following optical elements in combination with the flat mirror can form a real image?

- a) Convex lens
- b) Concave lens
- c) Concave mirror
- d) Convex mirror

Q4. Change in enthalpy (ΔH) for combustion of fuel is:

- a) Positive
- b) Zero
- c) Negative
- d) May be negative or positive

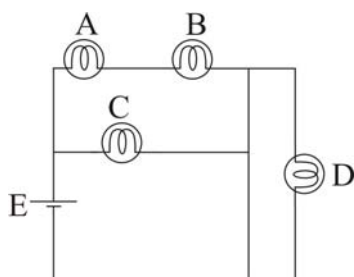
Q5. Phospholipid 'Cardiolipin' is unique to cell membranes of -

- a) Chloroplasts
- b) Mitochondria
- c) Golgi complex
- d) Nucleus

Q6. During aerobic respiration, energy is released in stepwise manner and ATP formation takes place with the help of this energy. What will happen if this energy is released at a single step instead of in parts?

- Incomplete oxidation of glucose takes place
- all the amount of energy can be utilized by the cell as more ATP molecules are produced
- maximum amount of this released energy is wasted in form of heat and cell may die
- cell will follow anaerobic pathway of respiration

Q7. Consider the circuit below:



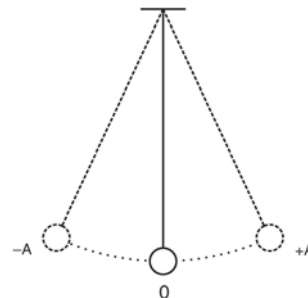
- If the bulb A burns out, then bulb C stays lighted, bulb B burns brightly
- If the bulb B burns out, then bulb C goes off and D stays lighted
- If the bulb C burns out, bulbs A and B go off
- If the bulb D burns out, the event is unnoticeable, and bulbs A, B, C stays lighted.

Q8. All molecules in the following set are allotropes of carbon:

- Charcoal, lead, coke
- Galena, glassy carbon, graphite
- Bucky balls, graphite, diamond
- Charcoal, wood, soot

Q9. A pendulum moves back and forth under the influence of gravity from $x = -A$ to $+A$ with time period T , as shown in figure. At time $t = 0$, it is at $x = +A$; when $T = 0.75T$, then:

- it is at $x = 0$ and travelling towards $x = +A$
- it is at $x = 0$ and travelling towards $x = -A$
- it is at $x = +A$ and is at rest
- it is between $x = 0$ and $+A$ and is travelling towards $x = -A$



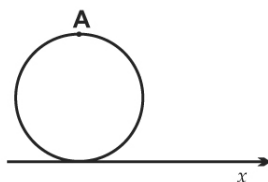
Q10. A solution of H_2SO_4 is 80% by weight, having specific gravity 1.73. Its normality is:

- a) 18.0
- b) 28.2
- c) 1.0
- d) 10.0

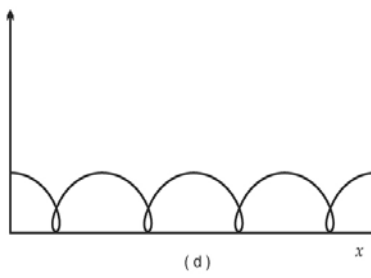
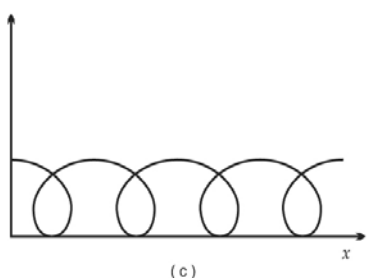
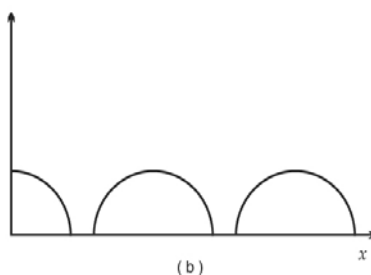
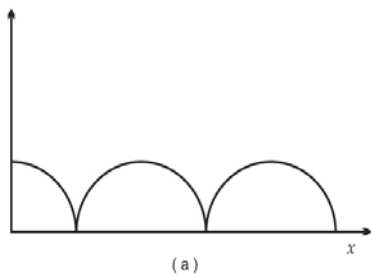
Q11. Which of the following complementary structures will be produced during replication of DNA sequence 5'-TpApGpAp-3' ?

- a) 5'-GpCpGpAp-3'
- b) 5'-ApTpCpTp-3'
- c) 5'-UpCpUpAp-3'
- d) 5'-TpCpTpAp-3'

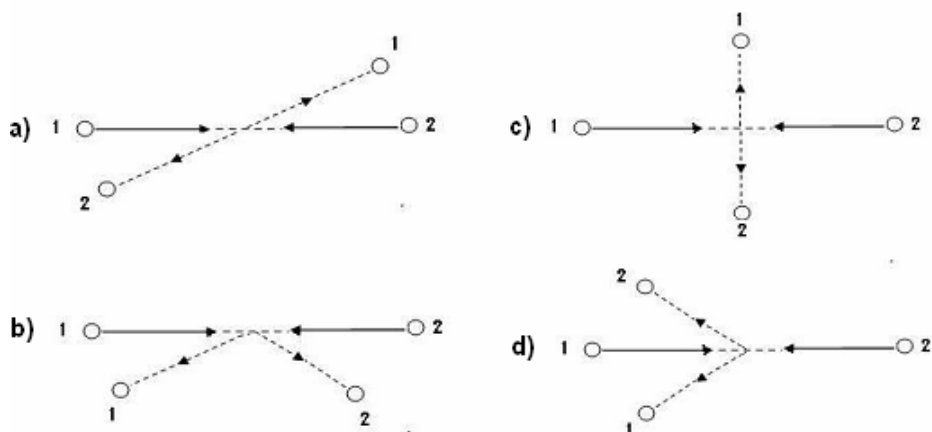
Q12. Let there be a rigid wheel rolling without sliding on a horizontal surface.



The path of point 'A' as seen by an observer on the ground, when the wheel is moving along x axis is:



Q13. Two identical balls (1 and 2) collide on a frictionless surface. Collision may or may not be elastic. Which of the views physically impossible situation?



Q14. Given 1 dm^3 of 0.15 M HCl and 1 dm^3 of 0.40 M HCl . What is the maximum volume of 0.25 M HCl which can be made by directly mixing the two solutions without adding water?

- 1 dm^3
- 2 dm^3
- 0.667 dm^3
- 1.667 dm^3

Q15. The inflammatory response is triggered whenever body tissues are injured. E.g. it occurs in response to physical trauma (a blow), intense heat and irritating chemicals as well as infection by viruses, fungi and bacteria.

Arrange the major events involved in the inflammatory process in the correct order as they occur in the body.

- Mast cells secrete histamine and cytokines.
- Phagocytosis by monocytes.
- Cytokines attract neutrophils and monocytes to injured site.
- Tissue injury
- Diapedesis or emigration of neutrophils & monocytes into the area through the capillary.
- Vasodilation of arterioles & increased capillary permeability causes redness and swelling.

- iv) \rightarrow i) \rightarrow iii) \rightarrow vi) \rightarrow v) \rightarrow ii)
- iv) \rightarrow i) \rightarrow v) \rightarrow vi) \rightarrow iii) \rightarrow ii)
- iv) \rightarrow vi) \rightarrow iii) \rightarrow ii) \rightarrow i) \rightarrow v)
- iv) \rightarrow i) \rightarrow vi) \rightarrow iii) \rightarrow v) \rightarrow ii)

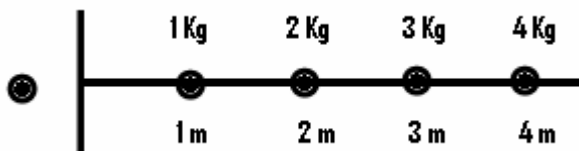
Q16. Baban arrives in your office for genetic counseling. Baban's brother Tapan dies at a young age from a hereditary Tay-Sach's disease. Both he and his sister Suman are unaffected. What is the chance of Suman being a carrier of this disease?

- a) $1/4$
- b) $2/3$
- c) $1/3$
- d) $1/9$

Q17. In the question above, if Baban's son Amar and Suman's daughter Asha are married to each other, and are expecting their first child. What is the chance that the child would have Tay-Sach's disease?

- a) $1/4$
- b) $1/9$
- c) $1/72$
- d) $1/36$

Q18. Four masses are located as shown in the figure. Acceleration due to gravity is same everywhere. What is the position of centre of gravity for the system?



- a) 2m
- b) 2.5m
- c) 3m
- d) 2.7m

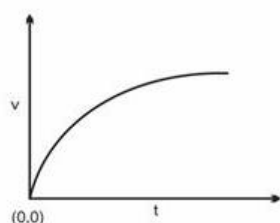
Q19. Calculate the mass of Lithium that contains same number of atoms as present in 8g of Magnesium. Atomic masses of lithium and magnesium are 7 and 24 respectively.

- a) 8g
- b) 3g
- c) 7g
- d) 2.3g

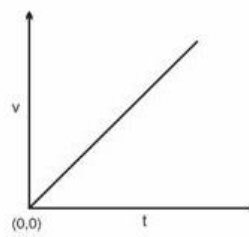
Q20. Which of the following are insoluble in water but will dissolve in aqueous NaOH?

- a) $\text{Mg}(\text{OH})_2$
- b) $\text{Fe}(\text{OH})_3$
- c) $\text{Cu}(\text{OH})_2$
- d) $\text{Zn}(\text{OH})_2$

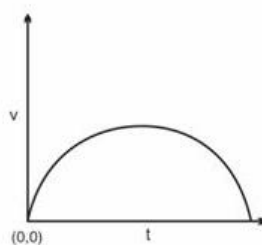
Q21. A large water tank is filled at a constant rate of 10 litres/min. It has an outlet of maximum flow of 10 litres/min at the bottom of the tank, but the output is proportional to the water present in the tank at any given time. How will the 'v', volume of water content in the tank, change with time?



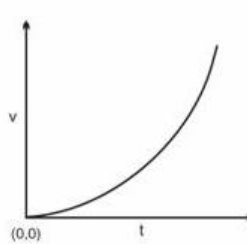
a)



b)



c)



d)

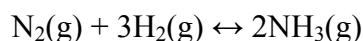
Q22. Human blood pressure (B.P.) is represented by what is called a "pressure picture". In a normal human being, the pressure picture is 120/80/40 where 120 represents systolic pressure, 80 represents diastolic pressure, and 40 represents the pulse pressure which is the resultant difference in the two pressures. In one of the patients, the pressure picture was found to be 140/110/30 suggesting that there has been –

- a) an increase in the resistance from arteries
- b) a decrease in the resistance from arteries
- c) an increase in the resistance from veins
- d) a decrease in the resistance from veins

Q23. Which of the following will be helpful in studying the details of surface texture of hair?

- a) Binocular microscope
- b) Scanning Electron Microscope
- c) Transmission Electron Microscope
- d) Phase- Contrast Microscope

Q24. Increase in pressure shifts the equilibrium of the reaction



- a) to produce more N_2
- b) to produce more NH_3
- c) to reduce the temperature of the reaction
- d) to produce more H_2

Q25. A solution containing $\text{Na}_2\text{CO}_3 + \text{NaOH}$ requires 300 ml of 0.1N HCl using phenolphthalein indicator. Methyl orange is then added to the above titrated solution and a further 25 ml of 0.2 N HCl is required. What is the amount of NaOH present in the solution?

- a) 0.8 g
- b) 1.0 g
- c) 1.5 g
- d) 2.0 g

Q26. When a ray of white light enters a prism, it begins to spread out in rainbow colours (Fig.1).



Fig.1

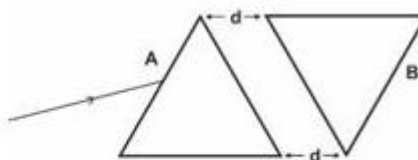


Fig.2

An inverted prism is brought close to this prism as shown in the Fig.2, Both the prisms are made of same material. If a ray of white light is incident on surface A and “d” is made zero then output from surface “B” will be:

- a) white light.
- b) rainbow colours which are converging.
- c) rainbow colours which are spreading out.
- d) no light comes out from surface B.

Q27. Which of the following oxides is an acid anhydride?

- a) Al_2O_3
- b) CO_2
- c) CO
- d) Red lead

Q28. In dispersive materials,

- a) the angle of refraction for a light ray depends on the wavelength of light
- b) the angle of refraction for a light ray does not depend on the wavelength of light
- c) the angle of reflection from the surface of the material does not depend on the wavelength of light
- d) Both (a) and (c) hold true.

Q29. There are two tracks A and B as shown in the figure. The direction of gravity is also shown in the figure.



If two similar balls begin to move at same uniform velocity at the same time which of the two balls will reach the end of the track faster?

- a) Ball on track A
- b) Ball on track B
- c) They will reach on the same time
- d) Cannot decide by the data given.

Q30. A human DNA molecule that was 6800\AA long was cut into 5 equal sized pieces using molecular scissors. How many nucleotides were present in each of the cut pieces?

- a) 800 nucleotides
- b) 400 nucleotides
- c) 80 nucleotides
- d) 1360 nucleotides

Q31. The relationship between respiration and photosynthesis is truly intricate. What will be the effect on the number of mitochondria and chloroplasts in a plant species occurring in higher altitude as compared to the same plant species in lower altitudes?

- a) number of mitochondria and number of chloroplasts will remain unchanged
- b) number of mitochondria in high altitude variety will be more
- c) number of chloroplasts in high altitude variety will be more
- d) both number of mitochondria and number of chloroplasts will be less in high altitude variety

Q32. A radioactive nuclide has a half life of 8 hours. Half life is the time in which 50 % of the nuclei disintegrate. The fraction of the nuclide which will disintegrate in 32 hours will be:

- a) 0.062
- b) 0.75
- c) 0.875
- d) 0.938

Q33. If length is increased by 20% and width is decreased by 20% then, which of the following is correct:

	Perimeter	Area
a)	Cannot say	Increases by 4%
b)	Increases by 4%	Cannot say
c)	Increases by 4%	Decreases by 4%
d)	Cannot say	Decreases by 4%

Q34. The alimentary canal of insect consists of mouth, pharynx, oesophagus, crop, proventriculus, gizzard and hindgut. In insects and birds, the gizzard helps in grinding the food. Select the statements that most accurately compare the gizzards of the bird and of insect.

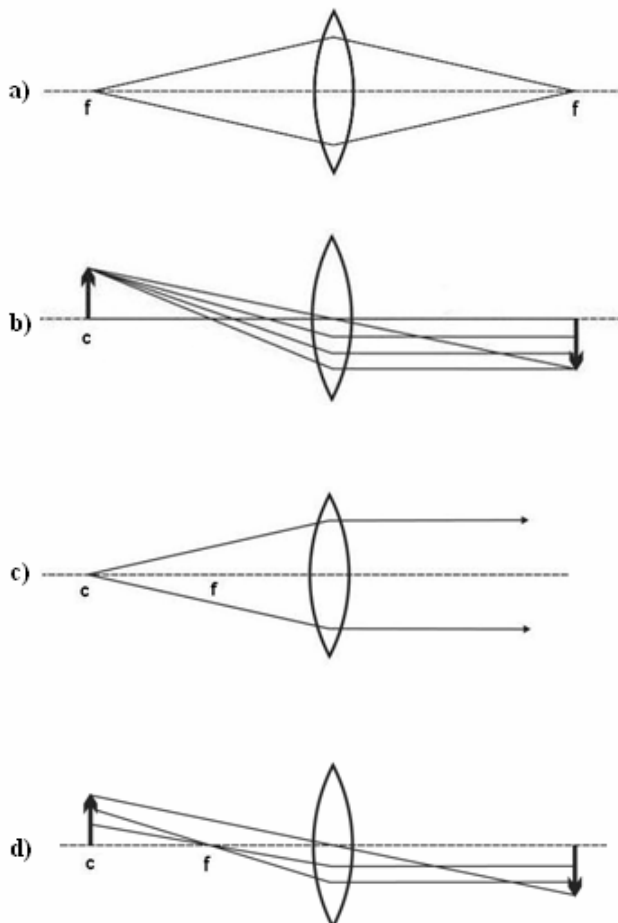
- i. The gizzard of the bird contains gravel and the gizzard of insect bears chitinous teeth.
- ii. The gizzard of the bird is muscular but that of the insect is devoid of muscles.
- iii. The gizzard in both cases is muscular.
- iv. The gizzard of bird bears bristles which are absent in the gizzard of insect.

- a) i and iii
- b) i and ii
- c) ii and iv
- d) iii and iv

Q35. In a room there are four objects – a wooden dish, steel dish, glass dish and a copper dish. If a fire is lit in the room so that it burns at 300 degrees centigrade, and is equidistant from all the four dishes, then after a long time the dishes can be listed in the increasing order of temperature. Which is the correct order of temperature of dishes?

- a) wooden dish > steel dish > glass dish > copper dish
- b) steel dish > copper dish > glass dish > wooden dish
- c) copper dish > steel dish > glass dish > wooden dish
- d) none of the above

Q36. Which of the following is correct? (Figures not to be scaled). The focal point of the lens is f and point c is equal to $2f$.



Q37. Occasionally a baby is born with a defect in septum of the heart. A tiny rupture, or a hole in the septum is a usual defect in such cases. If this kind of a defect exists in the inter-ventricular septum of a baby, what will be physiological consequences of this defect?

- Body receives deoxygenated blood, hence she may lose her life unless operated upon immediately.
- So long as baby is calm and quiet she remains normal, however on exertion or crying her skin turns bluish and baby gets exhausted.
- Heart cannot pump blood effectively, so blood does not reach all the tissues.
- Since valves are normal, systemic as well as pulmonary circulations are normal but blood pressure is low.

Q38. 50 cm³ of 1.5 M NaOH is titrated against 2M HCl. The pH of the reaction system after the addition of 35ml of HCl will be:

- a) 0.084 b) 1.23 c) 12.77 d) 7.95

Q39. The elements most likely to conduct electricity are the ones where electrons are not tightly bound to the nucleus. The trend for electrical conductivity in the following is:

- a) Li > Na > S > Ne
b) S > Li > Na > Ne
c) Ne > Na > Li > S
d) Na > Li > S > Ne

Q40. According to Darwin's theory of evolution, which of the following is a necessary condition (pre-requisite) for natural selection?

- a) variation among individuals of a population in some heritable characteristic
b) variation in habitat within the population of species
c) variation in environmental conditions that suit the individual
d) all of the above

Q41. If earth were to stop spinning then your weight on equator will be

- a) heavier by around 0.5%
b) heavier by around 5%
c) less by around 0.5 %
d) less by around 5%

Q42. A dwarf pea plant is treated with a hormone gibberellic acid. It was found that the plant height increases rapidly. Thus it was conformed that gibberellic acid brings about increase in the height of the plant. If this treated plant is crossed with normal dwarf pea plant, what kind of progeny would you expect?

- a) All tall
b) All dwarf
c) 50% tall and 50% dwarf
d) 75% tall and 25% dwarf

Q43. The volume occupied by 1.6g of oxygen under NTP conditions is:

- a) 1.12 dm³ b) 5.60 dm³ c) 11.20dm³ d) 22.4dm³

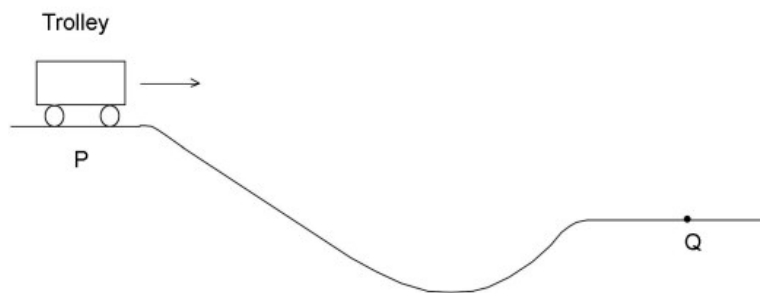
Q44. A perfectly spherical elastic balloon of 0.2 meter diameter was filled with hydrogen gas at sea level. What will be its volume at an altitude where pressure is 0.65 atm. What will be the diameter of the balloon?

- a) 0.20 m b) 0.23 m c) 0.10 m d) 0.32 m

Q45. A colourless crystalline solid 'B' dissolved easily in water. On addition of dilute HCl to the aqueous solution of 'B', no change was observed. When NaOH was added to the aqueous solution of 'B', a white ppt was obtained that dissolved in excess, giving a colourless solution. 'B' is:

- a) MgSO_4 b) $\text{Pb}(\text{NO}_3)_2$ c) AgNO_3 d) ZnSO_4

Q46. A trolley runs from point P to Q along a track, as shown in the figure. At point Q, its potential energy is 50 kJ less than at point P. At point P, the trolley has kinetic energy 5 kJ. Between P and Q, the work done against friction is 10 kJ. What is the kinetic energy at point Q?



- a) 35 kJ b) 45 kJ c) 55 kJ d) 65 kJ

Q47. Which of the following plant groups lack lignified cell walls in its cellular structure?

- a) Bryophytes b) Halophytes c) Pteridophytes d) Spermatophytes

Q48. Which of the following is not a simple harmonic motion?

- a) Earth moving around Sun
b) pendulum
c) oscillating mass attached to a spring
d) vibrating string of guitar

Q49. Gas liquid chromatography cannot be used to separate components which

- a) are volatile
- b) are non – volatile
- c) are liquids
- d) vaporize without decomposition

Q50. We know that the total amount of DNA from mitochondria and chloroplast will be one-third the quantity of nuclear DNA. It has been found that on an average the chloroplast DNA content is more than the mitochondrial DNA. This is because –

- a) mitochondria exhibit polyploidy only in meristems while chloroplasts show polyploidy even during cell maturation
- b) the average number of mitochondria in a plant are much less than the number of chloroplasts
- c) in general chloroplasts divide much more frequently than the mitochondria
- d) all of the above

Q51. Which of the following statement is true regarding the Kirchoff's laws?

- a) The junction rule is a statement of conservation of energy and the loop rule is a statement of conservation of charge
- b) The junction rule as well as the loop rule are statements of conservation of charge.
- c) The junction rule as well as the loop rule are statements of conservation of energy.
- d) The junction rule is a statement of conservation of charge and the loop rule is a statement of conservation of energy.

Q52. You have a large supply of light bulbs and a battery. You start with one light bulb connected to the battery and notice its brightness. You then add one light bulb at a time, each new bulb being added in series to the previous bulbs then

- a) Brightness of the bulbs will increase
- b) Current through the bulbs will increase
- c) Power transferred from the battery will decrease
- d) Lifetime of the battery will decrease

Q53. 0.1 mole of a weak acid HA and 0.1 mole of its sodium salt are dissolved in water and the solution is made upto 1000 cm^3 . If $K_a = 1.8 \times 10^{-5}$, the pH of the solution is:

- a) 2.4
- b) 3.70
- c) 4.74
- d) 5.00

Q54. Stem cells of an adult human being are said to be pluripotent because they have:

- a) potential to differentiate into cells of any type
- b) potential to reproduce indefinitely
- c) potential to differentiate into multiple but not all cell types
- d) all of the above

Q55. A ray of light is incident on the midpoint of a glass prism surface at an angle of 25.0° with the normal. For the glass, $n = 1.55$, and the angle of prism is 30.0° . What is the angle of incidence at the glass-to-air surface on the side opposite where the ray entered the prism?

- a) 14.2°
- b) 9.8°
- c) 19.2°
- d) light will not come out from the other surface

Q56. Which of the following has the smallest size?

- a) Ne b) O^{2-} c) F^- d) Na^+

Q57. A typical dicot possesses tap root system which has a primary root and many lateral roots and rootlets. Which of the following exhibits meristematic activity responsible for production of lateral roots?

- a) Vascular cambium
- b) Pericycle
- c) Endodermis
- d) Cork cambium

Q58. The ionization energy of nitrogen is more than that of oxygen because of:

- a) Greater attraction of nucleus
- b) Extra stability of half-filled p-orbitals
- c) Smaller size of nitrogen atom
- d) Poor shielding effect

Q59. A man of mass 100 kg stands on a wood plank of area 4 m^2 . What is the pressure exerted on the floor? Assume the area of a human foot to be 200 cm^2 .

- a) 500 N b) 25 N c) 50000 N d) 250 N

Q60. Human heart consisting of auricles and ventricles and muscular valves is an interesting functional system. The valve regulating the flow of impure blood from the right ventricle to the pulmonary artery is a:

- a) Tricuspid valve
 - b) Semilunar valve
 - c) Aortic valve
 - d) Bicuspid valve
-

Section B: Questions 61 to 68 are of 5 marks each. Marks will also be indicated in the questions if there are more than one part to it.

SECTION B (Long questions)

Q61. The human cerebrum is divided into right and left cerebral hemispheres. The outer 2-4mm of the cerebral hemispheres is known as the cerebral cortex. It consists of grey matter.

The left side of the cerebral cortex receives information from and controls movement of the right side of the body and vice-versa. A thick band of axons known as corpus callosum enables the right and left cerebral cortices to communicate. The left cerebral hemisphere is concerned with recognition of faces.

In a particular patient the corpus callosum was surgically removed as a treatment of last resort for the most extreme form of epilepsy in which fits occurred as often as 30 minutes. The patient appeared normal after recovery but exhibited a 'split-brain' effect.

Answer the following questions using options (Yes/No).

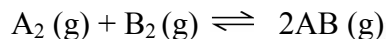
1. The patient is blindfolded and given a comb in the left hand. Then the blindfolded person is asked to locate the comb in a mixture of objects (mirror, ball, pen, etc) with his left hand. Will he succeed?
(1 mark)
2. If the doctor asks the blindfolded patient to say what the object was, would he be able to answer?
(1 mark)
3. The doctor asks the blindfolded patient to find the comb in the mixture of objects with his right hand. Would he be successful?

(1 mark)

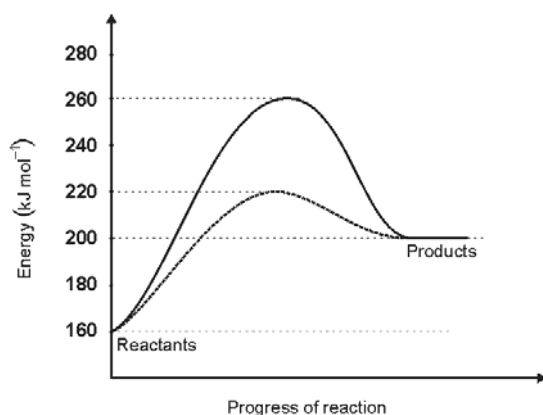
4. When the blindfolded patient is given the comb in right hand and asked to say what the object is, will he be able to answer? (1 mark)
5. If the patient is shown a photograph of a familiar face first in the left field of vision and then in the right field, would he be able to put a name to the face, in either field? (1 mark)

Total 5 marks

Q62. Consider the following hypothetical reaction



and answer the following:



- a) Calculate the energy of activation for the forward and backward reactions. (1 mark)
- b) Calculate ΔH (change in enthalpy) for the reaction. (1 mark)
- c) The dotted curve is the reaction path in presence of the catalyst. What is the lowering in activation energy in the presence of the catalyst? (1 mark)
- d) What will be the effect of increased pressure on equilibrium of the above reaction? (0.5 mark)

- e) What will happen if the temperature is increased (raised) by 10°C ? (0.5 mark)
- f) State with reason whether the position of equilibrium can be altered using catalyst? (1 mark)

Total 5 marks

Q63. a) A bus leaves a stop and accelerates at a constant rate for 5.0 seconds. During this time the bus travels 25 meters. After this the bus travels at a constant speed for 15 seconds. Then the driver notices a red light 18 meters ahead and applies brakes with acceleration ' a_b '. Assume that the bus decelerates at a constant rate and comes to a stop sometime later just at the light.

1. What was the initial acceleration of the bus?
 2. What was the velocity of the bus after 5 seconds?
 3. Calculate ' a_b '.
 4. How long did does the bus brake?
- (2 marks)

b) Two athletes Usha and Shiney are playing athletic games. Usha is running at a constant velocity towards Shiney who is stationary. When Usha is 12 meters away from Shiney, Shiney starts to accelerate at a constant rate of 1.5ms^{-2} .

1. What is the minimum velocity with which Usha needs to run in order to just catch up with Shiney?
 2. How long does Usha take to catch up with Shiney?
- (3 marks)

Total 5 marks

Q64. a) Solve:

$$\frac{3 \times 27^{x+1}}{3 \times 3^{3x+2}} + \frac{27 \times 3^{3x}}{1/3 \times 27^{x+1}} \quad (2\text{marks})$$

b) If $\tan x = b/a$,

write $\sqrt{\frac{a+b}{a-b}} + \sqrt{\frac{a-b}{a+b}}$ in terms of $\cos x$.

(3 marks)

Total 5 marks

Q65. a) A white inorganic solid 'A' on heating with dilute HCl gives a solution 'B' and gas 'C' with smell of rotten eggs. 'B' when treated with dilute NaOH gives a white ppt that dissolves in excess NaOH. Solution 'B' is soluble in dilute NH_4OH . On strong heating in air, 'A' gives a pungent gas 'D' and residue 'E' that dissolves in water. A dilute solution of 'E' gives a white ppt 'F' on treatment with BaCl_2 solution. Identify A, B, C, D, E, and F.

(3 marks)

b) A metal chloride 'X', on treatment with H_2S in acidic medium gives a black ppt. An aqueous solution of X when treated with a solution of SnCl_2 gives a white ppt turning grey in excess SnCl_2 . An aqueous solution of 'X' when treated with aqueous solution of KI gives a red ppt that dissolves in excess KI giving a colourless solution that is used for the detection of cation 'Y'. Identify 'X' and 'Y' with justification. and write all the reactions involved .

(2 marks)

Total 5 marks

Q66. a) There are two conducting plates A and B kept at a distance of 3 mm. Plate A is at +2 Volts and plate B is grounded (0 Volts). An electron begins to move from plate A with an initial velocity of ' v '. Find velocity ' v ' such that the electron travels up to plate B and when it reaches plate B its velocity is zero.

(2 marks)

b) A block of ice weighing 20 gm, at -10°C is mixed with 100 gm water at 10°C , in an insulated flask. What is the amount of water, in the flask, when equilibrium is reached?

(3 marks)

Total 5 marks

Q67. A water body characterized by nutrient rich water supports abundant growth of phytoplanktons and other water plants on its surface. Over time, the water body gets filled with a large number of such plants and the process is called eutrophication. In such water bodies, dissolved oxygen content is nil or very less.

With this information, answer the following questions:

1. Eutrophicated water usually looks turbid green in colour because of:
 - a) excessive growth of phytoplanktons
 - b) accumulation of minerals and nutrients
 - c) water body becoming polluted
 - d) release of degraded plant chlorophyll in water

(1 mark)
2. In eutrophicated water body, growth rate of phytoplanktons is high because of:
 - a) light penetrating the lower surfaces of water body
 - b) enrichment of nutrients in the water body
 - c) less of dissolved oxygen in the water body
 - d) aquatic fauna flourishes well in eutrophicated water body

(1 mark)
3. State whether the following statements are **True** or **False**
 - a) Mere increase of nutrients in water body leads to rigorous growth of benthic (plants rooted to the bottom with vegetation either submerged or emergent consisting of plants with their upper parts emerging from water) plants.

(0.5 mark)
 - b) Aquatic fauna thrives well in eutrophicated water body because there are plenty of phytoplanktons on which they could feed on.

(0.5 mark)
4. The reason for extremely low levels of dissolved oxygen in a eutrophicated water body is:
 - a) excessive growth of bacterial decomposers feeding on dead material which consume the dissolved oxygen
 - b) plants grow to such a great extent that they do not allow free flow of water hence oxygen in the atmosphere does not get mixed with water
 - c) the dissolved oxygen in water reacts with minerals and nutrients to form compounds and hence greatly reduces the quantity of oxygen
 - d) oxygen released is consumed by fishes in the eutrophicated water and hence the quantity of dissolved oxygen is very low

(1 mark)

5. Which of the following biotic organisms will not occur in eutrophicated water bodies?

- a) Phytoplanktons
- b) Zooplanktons
- c) Fishes
- d) Bacteria

(1 mark)

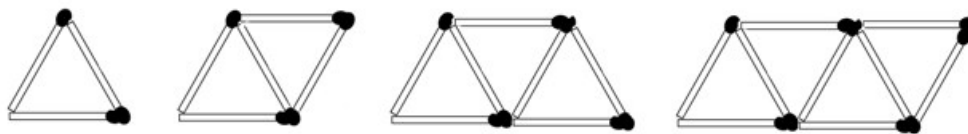
Total 5 marks

Q68. a) Find a positive integer solution to the following equation. Show your working.

$$x^2 + 615 = 2^n$$

(3 marks)

b) Examine the pattern of matchsticks



Find the match sticks required to make the a) 5th diagram, b) the nth diagram

(2 marks)

Total 5 marks